

# **Carbon Nanotubes as Catalyst Support**

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# Outline

- Background
- Physical Properties of Hyperion FIBRIL™ Nanotubes
- Functionalization of Carbon Nanotubes
- Preparation of Rigid Porous Carbon Nanotube Granules
- Application of FIBRIL Nanotubes as Catalyst Support
- Summary

# Background

## Activated Carbon as Catalyst Support

- Advantages:
  - Resistance to acid/basic media
  - Possibility to control porosity and surface chemistry
  - Chemical inertness
  - Easy recovery of precious metal
  - Large specific surface area, easy to obtain.
- Disadvantages:
  - Narrow microporosity (< 2nm)
  - Derived from natural resources → inconsistent quality, traces of impurities
  - Low mechanical and thermal stability

# Carbon Nanotubes(CNTs) as Catalyst Support

- Compared to activated carbons:
  - Mesoporosity (2-50nm) → improves mass transfer
  - High purity → avoids self-poisoning
  - Consistent material
  - High mechanical and thermal stability

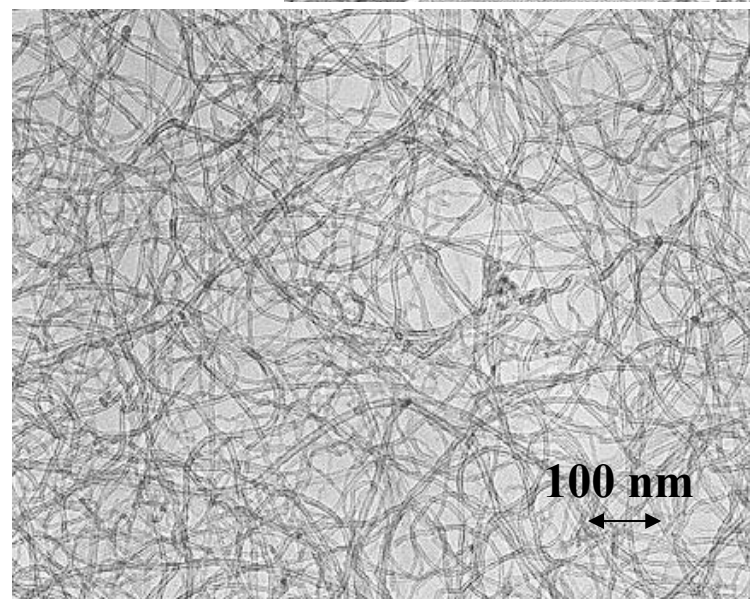
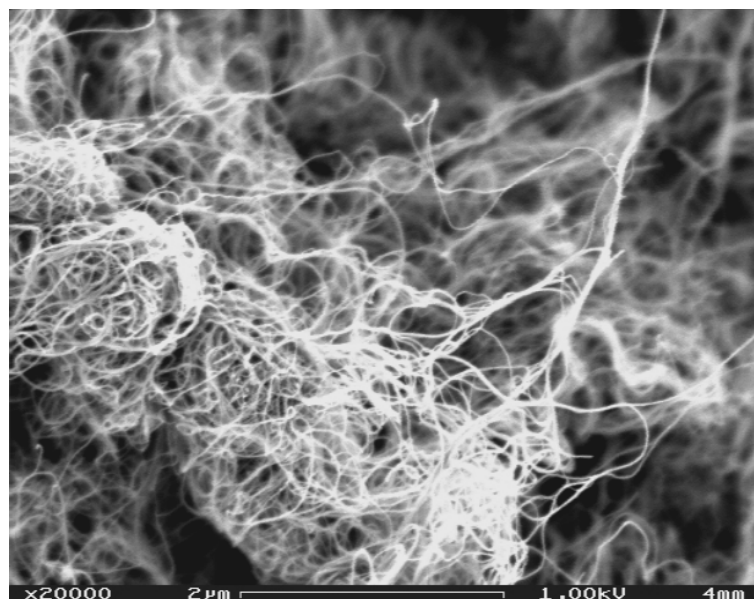
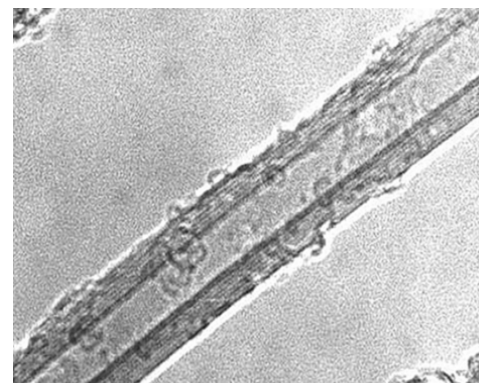
## CNTs as Catalyst Support (cont.)

- Challenges:
  - Controlling surface chemistry of CNTs → functionalization of CNTs
  - Increasing mechanical strength of CNT aggregates → preparation of rigid porous CNT granules
  - Maintaining the unique properties of CNTs, especially the mesoporosity

# Physical Properties of Hyperion FIBRIL Nanotubes

- Morphology of as-made Hyperion FIBRIL Nanotubes

OD: 8-15nm, Length: 1-100 $\mu$ m



- Porosity of as-made Hyperion FIBRIL Nanotubes

BET (m <sup>2</sup> /g)	Pore Volume (cm <sup>3</sup> /g)	
	Micro-	Meso-
240	0.03	2.01

Measured by N<sub>2</sub> adsorption at 77K

# Functionalization of CNTs

Oxidant	Nitric Acid	$(\text{NH}_4)_2\text{S}_2\text{O}_8/\text{H}_2\text{SO}_4$	$\text{H}_2\text{O}_2\text{-WO}_3$	$\text{H}_2\text{O}_2$
Titer (meqv/g)	1.01	0.76	0.38	0.16



# Preparation of Rigid Porous CNTs Granules

- Methods:
  - Direct extrusion of functionalized CNTs
  - Extrusion with addition of binder
- Crush Strength: 1/8" OD cylinder

Extrusion	Direct	With binder
Crush strength (lb/in)	38.9	61.3

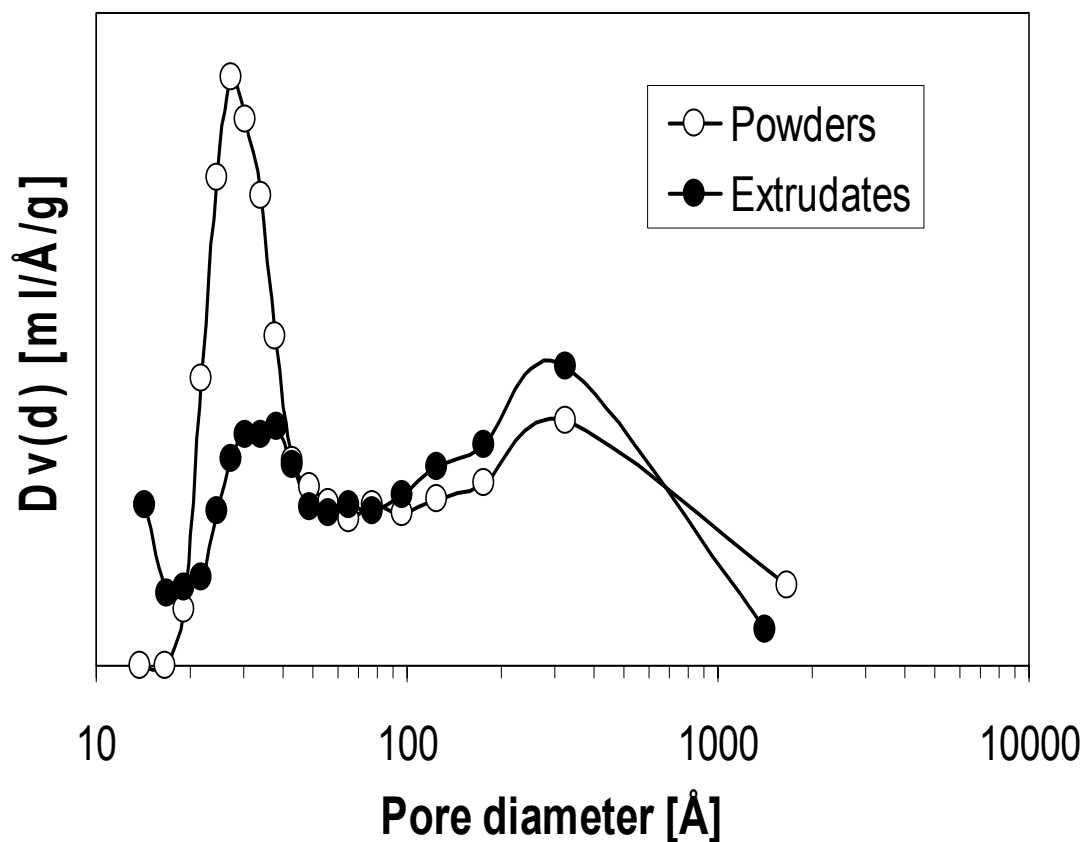
- Photograph of FIBRIL Nanotube Powders and Extrudates



- Pore Volume

Sample	Binder residue	Surface area	Micropore volume	Mesopore volume
	(wt%)	(m <sup>2</sup> /g)	(cm <sup>3</sup> /g)	(cm <sup>3</sup> /g)
Powders	0	240	0.03	2.01
Extrudates	1.7	244	0.01	1.40

- Porosity of Extrudates with Addition of Binder
  - Pore size distribution



# Application as Catalyst Support

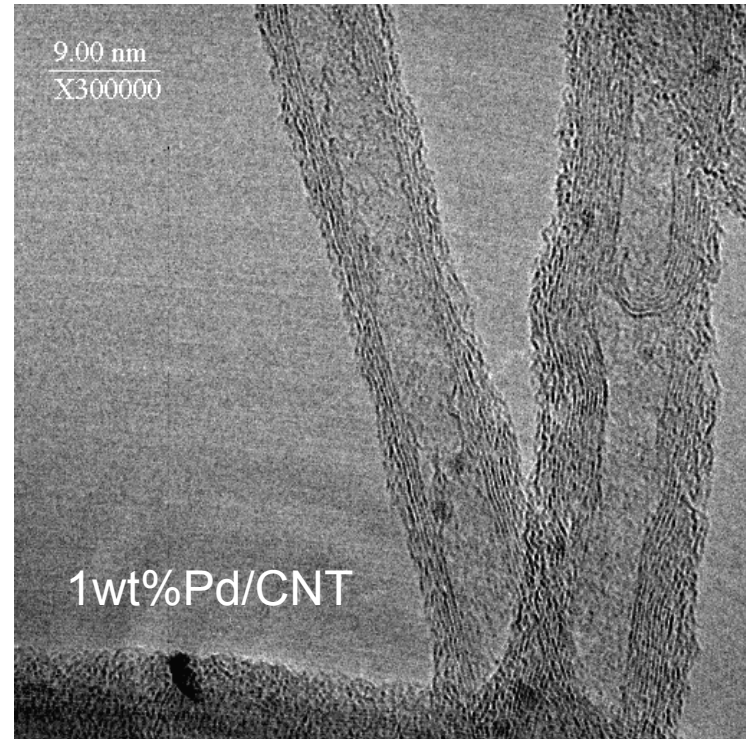
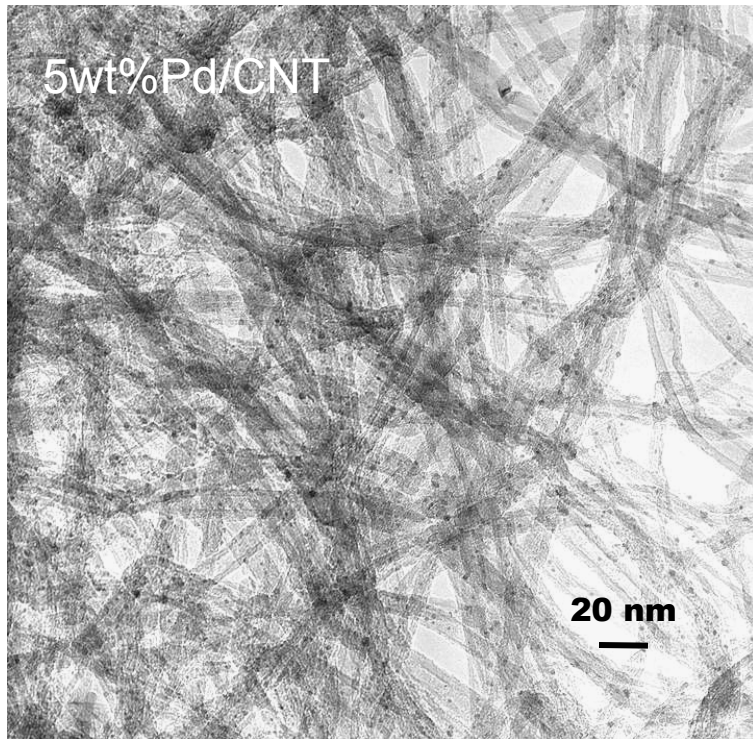
- CNT Supported Pd Catalysts<sup>#</sup>

Loading (wt%)	Dispersion* (%)	Particle size* (nm)
20	19.9	5.6
5	44.6	2.5
3	51.6	2.2
1	59.6	1.9
0.5	57.8	1.9
0.2	55.6	2.0

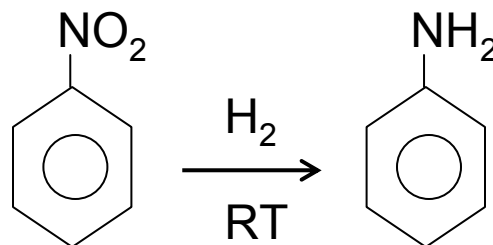
<sup>#</sup>: prepared by impregnation

<sup>\*</sup>: apparent Pd dispersions and particle size were measured by CO chemisorption at RT, and assuming CO:Pd=1

- TEM of Pd/CNT Catalysts



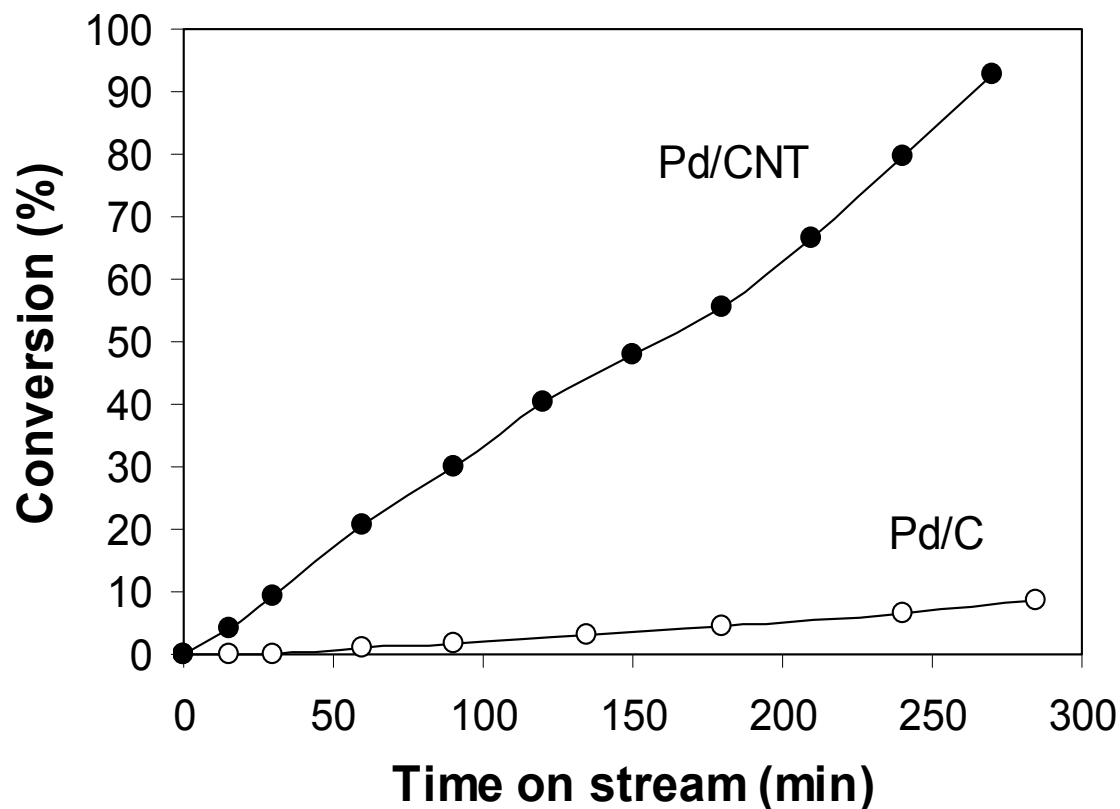
- Comparison of CNT with Activated Carbon as Support



(Gas-Liquid-Solid)

Support	Activated Carbon	CNT
Functionalization	HNO <sub>3</sub>	HNO <sub>3</sub>
Titer (meqv/g)	2.15	1.01
Pd Loading (wt%)	5	5
Pd Dispersion (%)	25.8	44.6
Pd Particle Size (nm)	4.3	2.5
Granule Size (μm)	150-300	150-300

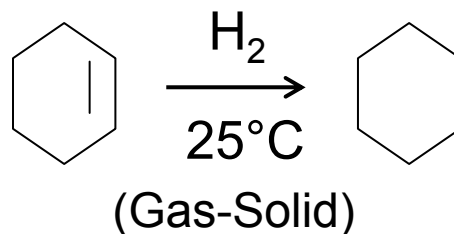
- Hydrogenation of Nitrobenzene



50mg of catalyst in 100ml of 5vol% nitrobenzene/2-propanol solution

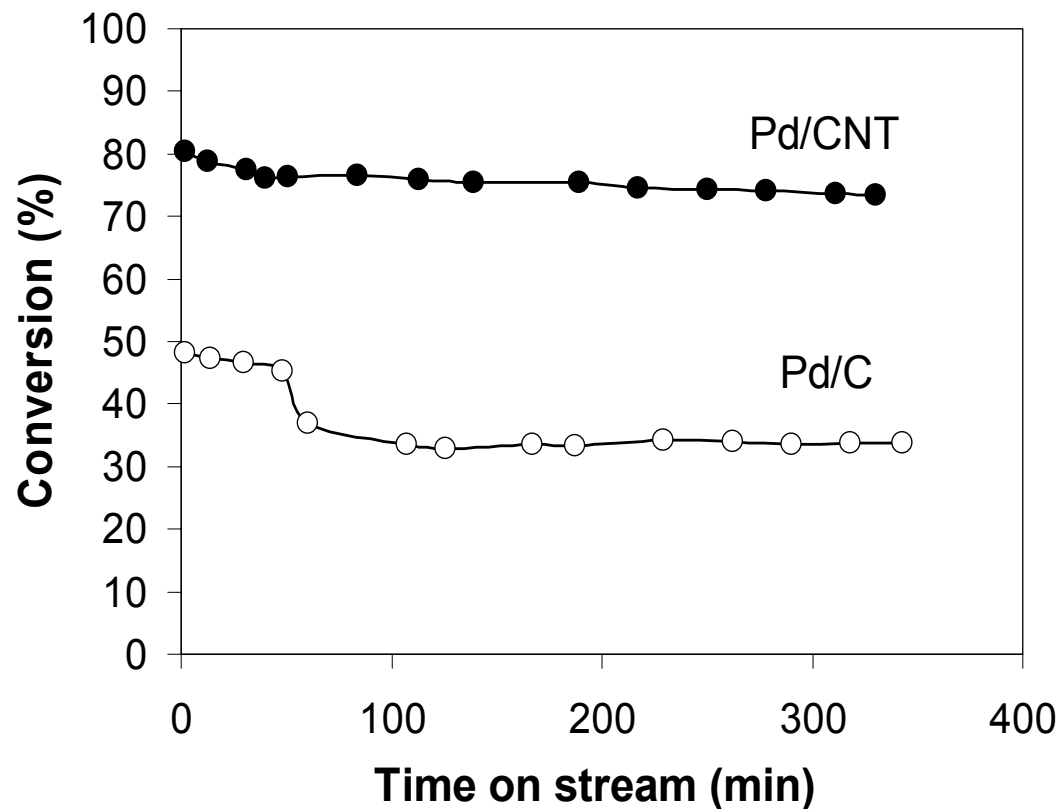


- Comparison of Pd/CNT with Commercial Pd/C Catalyst



Catalyst	Pd/C (PMC)	Pd/CNT
Appearance	Granules	Pellets
Pd Loading (wt%)	0.5	0.5
Pd Dispersion (%)	54.3	58.5
Pd Particle Size (nm)	2.1	1.9
Granule Size ( $\mu\text{m}$ )	420-840	420-840

- Hydrogenation of Cyclohexene



# Summary

- Successfully demonstrate . . .
  - Control of surface chemistry by CNT functionalization
  - Preparation of rigid porous CNT granules
  - Conservation of CNT mesoporosity in extrudates
- FIBRIL nanotubes are promising as catalyst support